Preparation of β^2 -Homotryptophan Derivatives for β -Peptide Synthesis¹)

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In view of the prominent role of the 1H-indol-3-yl side chain of tryptophan in peptides and proteins, it is important to have the appropriately protected homologs $H-\beta^2-HTrp-OH$ and $H-\beta^3-HTrp-OH$ (Fig.) available for incorporation in β -peptides. The β^2 -HTrp building block is especially important, because β^2 -amino acid residues cause β -peptide chains to fold to the unusual 12/10 helix or to a hairpin turn. The preparation of Fmoc – and $Z-\beta^2$ -HTrp(Boc) – OH by *Curtius* degradation (*Scheme 1*) of a succinic acid derivative is described (Schemes $2-4$). To this end, the (S)-4-isopropyl-3-[(N-Boc-indol-3-yl)propionyl]-1,3-oxazolidin-2-one enolate is alkylated with Br-CH₂CO₂Bn (Scheme 3). Subsequent hydrogenolysis, Curtius degradation, and removal of the Evans auxiliary group gives the desired derivatives of (R) -H $-\beta^2$ -HTrp $-OH$ (Scheme 4). Since the (R) form of the auxiliary is also available, access to (S) - β ²-HTrp-containing β -peptides is provided as well.

Introduction. - The tryptophan side chain, a $(1H\text{-}\text{indol-3-yl})$ methyl group, is hydrophobic and lipophilic, and, at the same time, it is capable of engaging in Hbonding. Of all amino acids, tryptophan has the highest affinity for the membranewater interface [2]. That makes tryptophan residues especially important for membrane proteins, where they are believed to serve as anchors on the periplasmic side of the membranes [3], and, indeed, membrane proteins have a significantly higher tryptophan content compared to soluble proteins. Also, antimicrobial peptides [4], containing several tryptophan residues, are widely distributed among living organisms and are believed to play important roles in the protection against potentially pathogenic microorganisms by disrupting the structural integrity of the microbial membranes. For example, indolicidin, which has antibiotic activity against bacteria [5], fungi [6], proteazoae [7], and viruses [8], is a cationic peptide amide consisting of only 13 amino acids, five of which are tryptophans; it also causes lysis of erythrocytes and is cytotoxic to human T lymphocytes [6]. Although tryptophan residues are localized at the membrane interface and are not inserted in the lipid bilayer [9], the peptide causes disruption of the cytoplasmic membrane [10], and the tryptophan residues appear to be essential for the hemolytic activity [11]. Another tryptophan-rich cationic peptide is tritrpticin, which possesses strong antimicrobial activity against bacteria and fungi [12]; it adopts four conformations in solution, but when it binds to membrane-mimetic sodium dodecyl sulfate micelles, only one major conformer is present [13]. Other peptides, such as lactoferricin B [14], mastoparan B [15], the 20-residue hybrid peptide

¹) Incorporation of a β^2 -HTrp residue into a turn-mimicking structure has been mentioned in a preliminary communication [1].

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 $CA(1-8)-MA(1-12)$ [16], and PMAP-23 [17] have been found to have antimicrobial activity, and the importance of at least part of their tryptophans for the observed activity indicates that the side chains of this amino acid function as anchors to the microbial membrane. Tryptophans are also essential for the formation of H_2O_2 by a variety of antibodies, which can participate in antibody-mediated cell killing [18].

Another important property of tryptophans is the fluorescence of the indole ring [19], which is highly sensitive to microenvironmental conditions and can, therefore, be used to study conformational changes of proteins and protein-membrane interactions: for instance, when tryptophan is inside a hydrophobic environment, a blue shift is observed³).

In view of the central role that tryptophan plays in peptides and proteins it is necessary to have access to the two homologs, 3-amino-4-(1H-indol-3-yl)butanoic acid $(H-\beta^3-HTrp-OH)$ and 3-amino-2-[(1H-indol-3-yl)methyl]propionic acid $(H-\beta^2-HTrp-OH)$ $HTrp-OH$) (*Fig.*), for incorporation into β -peptides. These analogs of α -peptides consist of homologated proteinogenic amino acids and have recently been shown to form two types of helices [21], turns [22] [23], and parallel as well as antiparallel sheets [22] [24] with short chain lengths, to be stable to peptidases [25], and to be able to mimic α -peptide turns [26], amphipathic helices [27], and cell-penetrating oligoarginines [28]. With β -peptides carrying tryptophan side chains, we have the means of mimicking yet other α -peptidic activities. Whereas the H $-\beta^3$ -HTrp $-OH$ is available by Arndt-Eistert homologation of the natural tryptophan [29], the β^2 -analog has not been described¹); on the other hand, β^2 -amino acid residues are especially important for the construction of turns [22] and of $12/10$ helices [21]. The preparation of β^2 -HTrp derivatives in either enantiomeric form is described herein.

Results and Discussion. – We chose to prepare Z-protected β^2 -HTrp 1a, suitable for solution-phase synthesis of β -peptides, and the Fmoc derivative 1b for solid-phase synthesis. The Boc group was applied for protection on the indole N-atom, since it is orthogonal to both Fmoc and Z protective groups. Moreover, Boc was expected to help diminish some side reactions known to occur with the Trp side chain during acidic

³⁾ Positively charged residues close to the benzene part or negative charges close to the pyrrole part of the indole ring cause λ_{max} shift to longer wavelengths (red shift) with the opposite configuration leading to a blue shift [20].

cleavage of peptides from $resins⁴$, a step in which the Boc group itself is readily removed.

We first tried to build the desired β -amino acid skeleton by diastereoselective aminomethylation [31] of the corresponding propionyl-oxazolidinone 2 [31a], a method we had previously applied for the preparation of other β^2 -amino acids, but experiments under various conditions were as fruitless as our preliminary attempts [31a]. We then tried to carry out the C,C-bond formation the other way around, *i.e.*, by indolylmethylation (with bromide 4 [32]) of 3-(3-aminopropionyl)-oxazolidinone derivatives $3a$ [31a] and $3b$, again without success⁵).

Another route for the attachment of a H_2NCH_2 group to a carboxylic acid (the Mannich transformation) is based on *Curtius* degradation of enantiomerically pure and regioselectively protected succinic acids, a route that can lead to either β^2 - or β^3 -amino acids [33] (Scheme 1). The precursors for the degradation can be obtained, for instance, by diastereoselective carboxymethylation of suitable chiral carboxylic acid derivatives [33a,b], by regio- and diastereoselective alkylation of chiral succinates [33c], or from malic acid [33d].

Preparation of the indolyl-substituted 3-propionyl-oxazolidinones 8a,b was straightforward: esterification of 3- $(1H$ -indol-3-yl)propionic acid (5) with CH₂N₂ [34], followed by Boc protection of the indole nucleus, provided the methyl indolepropionate 6 in 92% yield (Scheme 2). Hydrolysis of the ester function, followed by

⁴) During N-Boc deprotection with CF₃COOH the *t*-Bu moiety is removed, leaving the indole nucleus protected with a COOH group, thus preventing it from being alkylated, and from undergoing peptide reattachement or sulfonation. The COOH group is lost as $CO₂$ in H₂O to give the unprotected indole nucleus [30]. The Boc group could also protect the indole ring from being reduced to dihydroindole by silanes, which are often used as so-called scavengers in the removal/deprotection of peptides from resins [30b].

⁵) With a methylsulfonyl group at the indole N-atom (MeSO₂ instead of Boc in 4), a small amount (<10%) of alkylated product was detected in the reaction mixture.

Scheme 1. Curtius Degradation of Succinic Acid Derivatives for the Preparation of β^2 - or β^3 -Amino Acids

attachment of the oxazolidinone auxiliaries, led to the chiral derivatives **8a,b** in ca. 80% yield (Scheme 2). Since both enantiomeric forms of valine, the precursor of the oxazolidinones, are available, all enantiomers of the products reported herein are also accessible.

a) CH_2N_2 , Et_2O . b) Boc₂O, DMAP, MeCN. c) LiOH H_2O , THF/H₂O. d) Et₃N, t-BuCOCl. e) Auxiliary, LiCl.

For the carboxymethylation of indole-propionic acid derivatives 8 (Scheme 3), the benzyl bromoacetate was chosen, so that hydrogenolytic deprotection in the presence of the Boc group would be possible (under basic conditions, the auxiliary could be lost and/or partial epimerization of the newly formed stereogenic center could occur)⁶). In

⁶⁾ The trimethylsilylethyl ester of bromoacetic acid [35] was also used for the carboxymethylation of 8b. However, attempts to cleave the silylated ethyl ester group in the resulting product with Bu₄NF led to complete detachment of the auxiliary.

the optimization of the alkylation conditions, it was found that the NaHMDSgenerated Na enolate provided best yields and diastereoselectivities. Within experimental error, single diastereoisomers **9a,b** were present in the reaction mixture. Lithium and zinc enolates⁷) gave less-satisfactory results: the selectivity was excellent, but prolonged reaction times $(5 - 7 \text{ days})$ were necessary, which led to low conversion (36%) and required laborious separation from unreacted starting material 8.

Scheme 3. Preparation of Succinic Acid Derivatives 9 via Carboxymethylation of the N-Acyl-oxazolidinone 8

Contrary to our previous experience [31a], hydrogenolysis of the BnOCO group (in THF for 2 h) with the diphenyl-substituted oxazolidinone derivative 9b also led to the cleavage of $Ph_2C(5)-O$ bond in the auxiliary group⁸). This forced us to switch to the original Evans auxiliary with its less well-crystallizing derivatives. Thus, the Bn group of succinate 9a was cleaved by hydrogenolysis in dry THF with Pd/C as the catalyst (Scheme 4). From the resulting acid 10, an azide was generated in situ [33a] and decomposed with Curtius rearrangement in the presence of BnOH to give the Zprotected amino acid derivative 12. Removal of the auxiliary by $LiOH/H₂O₂$ afforded $Z-\beta^2$ -HTrp(Boc)–H (1a). Alcoholysis (BnOLi) of the intermediate 12 led to the benzyl ester **13** in 81% yield, which was converted to $Fmoc - \beta^2$ -HTrp(Boc)-H (**1b**) in 80% yield, by hydrogenolysis of both the ester Bn and the Z protecting groups, followed by Fmoc protection of the amino group [37]. Hydrogenolyses of succinate 9a in THF/MeOH 1:1 between 0° and room temperature produced a mixture of epimeric indoline derivatives 11; although these could be used for the *Curtius* rearrangement, followed by re-oxidation with DDQ, the overall yield was poorer.

Thus, we have developed a useful route from commercial 3-(1H-indol-3-yl)propionic acid to the desired N-Fmoc- and N-Z-protected β^2 -homotryptophans for β peptide synthesis⁹). It is remarkable that the preparation of compounds as simple as the amino-acid derivatives 1 in enantiomerically pure form takes nine (for 1a) and eleven steps (for 1b) from commercial precursors, using an auxiliary approach.

⁷⁾ ``Zinc enolates' were generated by addition of ZnCl_2 to the solutions of the lithium enolates.
⁸) Hydrogenolysis of the Ph₂C(5)–O bond of 4-substituted 5,5-diphenyloxazolidin-2-ones has Hydrogenolysis of the Ph₂C(5)-O bond of 4-substituted 5,5-diphenyloxazolidin-2-ones has been used to prepare chiral, diphenylmethyl-substituted amines [36].

⁹) The enantiomer purity of 1a and 1b was not determined, but we know that β -peptides prepared with incorporation of β -HTrp (from **1a**) are diastereoisomerically pure [1].

a) H_2 (1 atm), 10% Pd/C, THF. b) H_2 (1 atom), 10% Pd/C, THF/MeOH 1:1. c) (PhO)₂P(O)N₃, Et₃N, BnOH, toluene, reflux. d) DDQ, CH₂Cl₂. e) LiOH \cdot H₂O, 30% H₂O₂, THF/H₂O. f) BnOH, BuLi, THF. g) H₂, 10% Pd/ C, EtOH. h) Fmoc-OSu, Na_2CO_3 , acetone/H₂O. DDQ = 2,3-Dichloro-5,6-dicyano-1,4-benzoquinone, Su = Nsuccinimidyl.

Experimental Part

1. General. Abbreviations: Boc₂O: di(tert-butyl) dicarbonate, DCTB: 2-{(2E)-3-[4-(tert-butyl)phenyl]-2methylprop-2-enylidene}malonitrile, DDQ: 2,3-dichloro-5,6-dicyano-1,4-benzoquinone, DHB: 2,5-dihydroxybenzoic acid, DMAP: 4-dimethylaminopyridine, FC: flash chromatography, h.v.: high vacuum, 0.01 - 0.1 Torr, NaHMDS: sodium hexamethyldisilazan. Solvents for chromatography and workup procedures were distilled from Sikkon (anh. CaSO₄; Fluka); Et₂O was distilled from KOH/FeSO₄. Et₃N was distilled from CaH₂ and stored over molecular sieves (4 Å) . Pivaloyl chloride was distilled and stored under Ar. LiCl and ZnCl₂ were dried under h.v. at 200 $^{\circ}$ for 24 h. All other reagents were used as received from *Fluka*. The methyl 3-(1H-indol-3-yl)propionate was prepared according to the procedure in [34]. TLC: Merck silica gel 60 F_{254} plates; detection with UV and anisaldehyde or I₂. FC: Fluka silica gel 60 (40-63 µm); at ca. 0.5 bar. M.p.: Büchi 510 apparatus; uncorrected. Optical rotations: Perkin-Elmer 241 polarimeter (10 cm, 1-ml cell) at r.t. IR Spectra: Perkin-Elmer 782 spectrophotometer. NMR Spectra: *Bruker AMX-500* (¹H: 500 MHz, ¹³C: 125 MHz), *AMX-400* (¹H: 400 MHz, 13 C: 100 MHz); chemical shifts δ in ppm downfield from internal Me₄Si (=0 ppm); *J* values in Hz; some compounds show the presence of rotamers, which are indicated. MS: IonSpec Ultima-Maldi FT/ICR mass spectrometer, matrix DHB or DCTB, 4.7 T, N₂ laser at 337 nm. Elemental analyses were performed by the Microanalytical Laboratory of the Laboratorium für Organische Chemie, ETH-Zürich.

(R)-3-{[(Benzyloxy)carbonyl]amino}-2-({1-[(tert-butoxy)carbonyl]-1H-indol-3-yl}-methyl)propionic Acid (1a). To a stirred soln. of amide 12 (627 mg, 1.108 mmol) in THF (16 ml), H_2O_2 (30% aq. sol.; 0.45 ml, 4.37 mmol) and soln. of LiOH \cdot H₂O in 5.6 ml H₂O were added at 0°. The mixture was stirred at 0° for 2.5 h before addition of a sat. aq. soln. of Na₂SO₃ (5 ml). The soln. was partially conc. in vacuo, diluted with H₂O and acidified to pH 1 – 2 with 10% HCl at 0°. A white solid precipitated, was extracted with EtOAc (2 \times), the org. layer was dried ($MgSO₄$) and concentrated in vacuo to yield 639 mg of an oil. FC (pentane/Et₂O/AcOH $100:100:1$) yields **1a** (494 mg, 98%). White foam. M.p. $63-65^{\circ}$. [α] $_{1D}^{1\text{rt}} = +0.57$ ($c = 0.6$, CHCl₃). IR (CHCl₃): 3448w, 2982w, 1724s, 1514m, 1453m, 1371m, 1157m, 1092m, 1040w, 1020w, 855w. ¹H-NMR (400 MHz, CDCl₃; signals of rotamers in italics): 1.66 (s, t-Bu); $2.81 - 2.89$, $2.90 - 2.94$ (m, 1 H, indCH₂); 3.0 - 3.15 (m, 2 H, indCH₂). $C(O)CH$; 3.2 – 3.35, 3.4 – 3.5 (m, 1 H, NCH₂); 3.5 – 3.6 (m, 1 H, NCH₂); 5.04 – 5.11 (m, OCH₂); 5.22 – 5.30, 6.6 – 6.7 (m, NH); 7.13 - 7.52 (m, 9 arom. H); 8.12 (br. s, arom. NCH). ¹³C-NMR (100 MHz, CDCl₃): 24.9, 25.2 (CH₂); 28.2 (Me); 41.8, 42.3 (CH₂); 45.3, 45.6 (CH); 66.9, 67.3 (CH₂); 83.7 (C); 115.4 (CH); 116.7 (C); 118.7, 122.6, 123.7, 123.9, 124.5, 127.8, 128.2, 128.4, 128.5 (CH); 130.2, 135.5, 136.3, 149.7, 156.5, 157.8, 177.5, 178.4 (C). HR-MALDI-MS: 475.1845 (+1.05 ppm) $(C_2,H_{28}N_2O_6Na^+$; calc. 475.1840). Anal. calc. for $C_2,H_{28}N_2O_6$ (452.50): C 66.36, H 6.24, N 6.19; found: C 66.00, H 6.19, N 6.09.

(R)-3-({[(9H-Fluoren-9-yl)methoxy]carbonyl}amino)-2-({1-[(tert-butoxy)carbonyl]-1H-indol-3-yl}methyl)propionic Acid (1b). To a stirred soln. of BnOH (0.15 ml, 1.45 mmol, 2.36 equiv.) in THF (6 ml) at -78° , BuLi (1.6m in hexane, 0.6 ml, 1.56 equiv.) was added, followed by a soln. of amide 12 (347 mg, 0.613 mmol, 1 equiv.) in 2 ml of THF. After 5 min, the temp. was allowed to rise to 0° , and stirring was continued for 3 h before NH₄Cl soln. was added, and the mixture was diluted with H_2O and Et₂O. The org. phase was separated, washed with brine, dried (MgSO₄), and concentrated in vacuo. FC (pentane/Et₂O 3:1 to 2:1) yielded the benzyl ester 13 as a colorless oil (269 mg, 81%). Part of this oil (157 mg, 0.289 mmol) was dissolved in EtOH (5 ml) and stirred under H₂ atmosphere (1 atm, balloon) with 10% Pd/C (15 mg) for 2 h at r.t. Pd/C was filtered off, washed with EtOH, and the filtrate was concentrated in vacuo to yield the crude amino acid, which was suspended in 0.15 aq. Na₂CO₃ (3.9 ml, 0.585 mmol, 2 equiv.). The suspension was treated with a soln. of Fmoc-OSu (117 mg, 0.347 mmol, 1.2 equiv.) in acetone (3.5 ml); all solids dissolved. The soln. was stirred for 21 h at r.t., the solvent was partially evaporated in vacuo and diluted with H₂O (100 ml) and Et₂O (50 ml). The org. phase was washed with 0.15 Ma₂CO₃ (35 ml). Combined H₂O phases were acidified to pH 2 - 3 with 1M HCl and extracted with AcOEt $(2 \times 60 \text{ ml})$. AcOEt phases were dried (MgSO₄) and concentrated in vacuo to yield 158 mg of colorless solid. Recrystallization (cyclohexane/AcOEt) yielded 1b (125 mg, 80% from 13). Colorless solid. M.p. 125 128° . [α]_i₁t. = -5.0 (c = 1.04, CHCl₃). IR (CHCl₃): 3451w, 3007w, 1723s, 1516m, 1452m, 1371m, 1156m, 1090m, $1019w$, $857w$. ¹H-NMR (400 MHz, CD₃OD; signals of rotamers in italics): 1.57, 1.64 (s, t-Bu); 2.5, 2.7, 2.85 – 3.05, $3.25 \ (m, \text{indole-CH}_2, \text{C(O)CH})$; $3.94, 4.19 \ (t, J = 6.8, \text{CHCH}_2\text{O})$; $4.32 - 4.34 \ (m, \text{CHCH}_2\text{O})$; $7.17 - 7.40 \ (m, \text{H} \cdot \text{O})$ 6 arom. H); 7.47 (s, 1 arom. H); 7.55 (d, $J = 7.7$, 1 arom. H); 7.63 (d, $J = 7.4$, 2 arom. H); 7.70 – 7.80 (m, 2 arom. H); 8.00, 8.07 (br. d, J = 8.1, arom. NCH). ¹³C-NMR (100 MHz, CD₃OD): 26.0 (CH₂); 28.4 (Me); 43.6, 47.3 (CH); 67.8 (CH₂); 84.7 (C); 116.1 (CH); 119.3 (C); 120.1, 120.9, 123.6, 124.6, 125.4, 126.3, 128.2, 128.8 (CH); 131.8, 136.9, 142.6, 145.4, 151.1, 158.9, 177.6 (C). HR-MALDI-MS: 563.2465 (+55 ppm) $(C_{32}H_{32}N_{2}O_{6}Na^{+}$; calc. 563.2153). Anal. calc. for C₃₂H₃₂N₂O₆ \cdot 1/2 H₂O (549.62): C 69.93, H 6.05, N 5.10; found: C 69.77, H 6.18, N 5.16.

Methyl 3-{1-] (tert-Butoxy)carbonyl]-1H-indol-3-yl]propionate (6). A stirred soln. of methyl 3-{1H-indol-3-yl)propionate [34] (1.233 g, 6.018 mmol, 1 equiv.) in MeCN (10 ml) was treated with a soln. of Boc₂O (1.379 g, 6.318mmol, 1.05 equiv.) in 2 ml of MeCN and with DMAP (40 mg, 0.327 mmol, 0.05 equiv.). After stirring at r.t. for 1 h, additional Boc₂O (156 mg, 0.715 mmol, 0.1 equiv.) was added, and stirring was continued for 90 min. The mixture was concentrated in vacuo, diluted with AcOEt (40 ml), washed with 1M HCl $(2 \times 30 \text{ ml})$ and brine (30 ml) , dried $(MgSO₄)$, and concentrated in vacuo. FC (pentane/Et₂O 10:1) yielded 6 (1.678 g, 92%). Colorless oil. IR (CHCl₃): 3005w, 2982w, 1729s, 1477w, 1453m, 1371m, 1309w, 1086m, 1019w, 856w. ¹H-NMR $(400 \text{ MHz}, \text{CDCl}_3)$: 1.67 (s, t-Bu); 2.70 – 2.75 (m, CH₂); 3.01 – 3.06 (m, CH₂); 3.70 (s, OMe); 7.24 (ddd, J = 1.1, 7.2, 7.7, 1 arom. H); 7.29 – 7.34 $(m, 1 \text{ arom. H})$; 7.38 $(s, 1 \text{ arom. H})$; 7.52 $(ddd, J=0.8, 1.3, 7.7, 1 \text{ arom. H})$; 8.11 (br. $d, J = 7.7$, arom. NCH). ¹³C-NMR (100 MHz, CDCl₃): 20.4 (CH₂); 28.2 (Me); 33.8 (CH₂); 51.7 (Me); 83.5 (C); 115.3, 118.8 (CH); 119.4 (C); 122.4, 122.5, 124.4 (CH); 130.3, 135.5, 149.8, 173.4 (C). HR-MALDI-MS: 326.1360 (-0.92 ppm) ($C_{17}H_{21}NO_4Na^+$; calc. 326.1363). Anal. calc. for $C_{17}H_{21}NO_4$ (303.35): C 67.31, H 6.98, N 4.62; found: C 67.53, H 7.00, N 4.45.

3-{1-[(tert-Butoxy)carbonyl]-1H-indol-3-yl}propionic Acid (7). A stirred soln. of 6 (1.575 g, 5.192 mmol) in THF (15 ml) was diluted with H₂O (5 ml) and treated with LiOH \cdot H₂O (500 mg, 11.916 mmol, 2.3 equiv.). After stirring at r.t. for 9 h, most of the THF was removed in vacuo, and the residue was acidified to pH $2-3$ by 10% soln. of tartaric acid and extracted with CH₂Cl₂ (3×25 ml). Each org. phase was washed with H₂O (25 ml). The combined org. phases were dried (MgSO₄) and concentrated in vacuo to yield 7 (1.406 g, 94%) of colorless oil, which solidified while standing. Crude product was pure by ¹ H-NMR spectrum and was used in the next reaction without further purification. FC (pentane/Et₂O/AcOH 2:1:0.02) and recrystallization from hexane/ AcOEt of small sample yielded anal. pure 7. White solid. M.p. 117°. IR (CHCl3): 3008w, 2981m, 2932w, 1725s, 1475w, 1453s, 1371s, 1340w, 1309w, 1086s, 1043w, 1020w . ¹ H-NMR (400 MHz, CDCl3): 1.66 (s, t-Bu); 2.78 $(m, CH₂)$; 3.04 $(m, CH₂)$; 7.24 $(dd, J=1.1, 7.3, 7.7, 1$ arom. H); 7.32 $(dd, J=1.2, 7.2, 8.3, 1$ arom. H); 7.41 $(s, 1 \text{ atom. } H)$; 7.52 (ddd, J = 0.7, 1.3, 7.7, 1 arom. H); 8.11 (br. d, J = 7.5, arom. NCH). ¹³C-NMR (100 MHz, CDCl₃): 20.1 (CH₂); 28.2 (Me); 33.6 (CH₂); 83.6 (C); 115.3, 118.7 (CH); 119.1 (C); 122.5, 122.6, 124.5 (CH), 130.2, 135.5, 149.8, 178.7 (C). HR-MALDI-MS: 312.1205 (-0.32 ppm) (C₁₆H₁₉NO₄Na⁺; calc. 312.1206). Anal. calc. for $C_{16}H_{19}NO_4$ (289.33): C 66.42, H 6.62, N 4.84; found: C 66.39, H 6.59, N 4.84.

(S)-3-(3-{1-[(tert-Butoxy)carbonyl]-1H-indol-3-yl}-1-oxopropyl)-4-isopropyloxazolidin-2-one (8a). To a soln. of 7 (6.106 g, 21.104 mmol, 1.05 equiv.) in THF (100 ml), Et₃N (7.3 ml, 52.37 mmol, 2.6 equiv.) and pivaloyl chloride (2.60 ml, 21.13 mmol, 1.05 equiv.) were added at -30° . The resulting white suspension was stirred at -30° for 90 min; LiCl (980 mg, 23.12 mmol, 1.15 equiv.) and (S)-4-isopropyloxazolidin-2-one (2.596 g, 20.099 mmol, 1 equiv.) were added, and the mixture was allowed to warm slowly to r.t. over 11 h. The mixture was treated with sat. NH₄Cl (50 ml), diluted with H₂O (100 ml) and Et₂O (250 ml). The org. phase was separated, washed with 1M HCl $(2 \times 100 \text{ ml})$, 1M NaOH $(2 \times 100 \text{ ml})$, and brine (100 ml). Each H₂O phase was re-extracted with Et₂O (250 ml). Combined org. phases were dried $(MgSO₄)$ and concentrated in vacuo. FC (pentane/Et₂O 2:1) yielded **8a** (6.675 g, 83%). White solid. M.p. 44° . α ₁th= +48.0 (c = 0.9, CHCl₃). IR (CHCl3): 3032w, 2972m, 2877w, 1780s, 1724s, 1608w, 1486w, 1453s, 1387s, 1308m, 1086s, 1032m, 1019m. ¹ H-NMR $(400 \text{ MHz}, \text{CDCl}_3)$: 0.85 $(d, J = 6.9, \text{ Me})$; 0.91 $(d, J = 7.0, \text{ Me})$; 1.66 $(s, t$ -Bu); 2.31 – 2.43 (m, Me_2CH) ; 3.01 – 3.13 $(m, CH₂)$; 3.29 (ddd, J = 7.1, 7.8, 17.0, 1 H, CH₂); 3.41 (ddd, J = 6.8, 8.4, 17.0, 1 H, CH₂); 4.20 (dd, J = 3.3, 9.1, 1 H, OCH₂); 4.25 (dd, J = 8.1, 9.1, 1 H, OCH₂); 4.44 (ddd, J = 3.3, 3.9, 8.1, NCH); 7.24 (ddd, J = 1.1, 7.2, 7.7, 1 arom. H); 7.28 – 7.33 $(m, 1 \text{ atom. H})$; 7.42 $(s, 1 \text{ atom. H})$; 7.58 $(ddd, J = 0.7, 1.3, 7.7, 1 \text{ atom. H})$; 8.12 $(\text{br. } d, J = 0.7, 1.3, 7.7, 1.4)$ 7.8, arom. NCH). 13C-NMR (100 MHz, CDCl3): 14.7, 18.0 (Me); 19.8 (CH2); 28.2 (Me); 28.4 (CH); 35.3 (CH2); 58.5 (CH); 63.5 (CH2); 83.4 (C); 115.2, 119.0 (CH); 119.4 (C); 122.4, 122.9, 124.3 (CH); 130.3, 135.8, 149.7, 154.1, 172.4 (C). HR-MALDI-MS: 423.1891 (+0.24 ppm) $(C_2H_{28}N_2O_5Na^+$; calc. 423.1890). Anal. calc. for $C_{22}H_{28}N_2O_5$ (400.47): C 65.98, H 7.05, N 7.00; found: C 65.97, H 7.07, N 7.15.

(S)-3-(3-{1-[(tert-Butoxy)carbonyl]-1H-indol-3-yl}-1-oxopropyl)-4-isopropyl-5,5-diphenyloxazolidin-2 one (8b). Analogously to the preparation of $8a$, 7 (308 mg, 1.064 mmol) and (S)-4-isopropyl-5,5-diphenyloxazolidin-2-one (285 mg, 1.013 mmol) were reacted. FC (pentane/Et₂O 8:1) yielded **8b** (459 mg, 82%). White solid. M.p. 78–80°. $[a]_D^{\text{rt}} = -134.9$ (c = 0.77, CHCl₃). IR (CHCl₃): 3008w, 1781s, 1724s, 1494w, 1452s, 1371s, $1320w$, $1087m$, $1048w$, $1019w$, $991w$, $859w$. 1 H-NMR (500 MHz, CDCl₃): 0.75 (d, $J = 6.8$, Me); 0.87 (d, $J = 7.0$, Me); 1.65 (s, t-Bu); 1.94 – 2.01 (m, Me₂CH); 2.89 – 2.95 (m, 1 H, CH₂); 2.99 – 3.05 (m, 1 H, CH₂); 3.11 (ddd, J = 5.8, 9.2, 16.4, 1 H, CH₂); 3.35 (ddd, J = 6.1, 9.5, 16.4, 1 H, CH₂); 5.40 (d, J = 3.4, NCH); 7.19 – 7.40 $(m, 11 \text{ arom. H})$; 7.44 – 7.48 $(m, 2 \text{ arom. H})$; 7.53 $(dd, J = 0.7, 1.2, 7.8, 1 \text{ arom. H})$; 8.1 (br. $d, J = 6.5$, arom. NCH). ¹³C-NMR (125 MHz, CDCl₃): 16.4 (Me); 20.0 (CH₂); 21.7, 28.2 (Me); 29.9 (CH); 35.0 (CH₂); 64.6 (CH); 83.3, 89.5 (C); 115.2, 119.0 (CH); 119.3 (C); 122.4, 122.8, 124.3, 125.6, 125.9, 126.6, 128.0, 128.4, 128.6, 128.9 (CH); 130.3, 135.5, 138.1, 142.3, 149.7, 153.0, 172.3 (C). HR-MALDI-MS: 575.2522 (+1.04 ppm) $(C_{34}H_{36}N_2O_5Na^+$; calc. 575.2516). Anal. calc. for $C_{34}H_{36}N_2O_5$ (552.66): C 73.89, H 6.57, N 5.07; found: C 74.00, H 6.78, N 5.03.

Benzyl (R)-3-({1-[(tert-Butoxy)carbonyl]-1H-indol-3-yl}methyl)-4-[(S)-4-isopropyl-2-oxooxazolidin-3 yl)-4-oxobutanoate (9a). NaHMDS (2m in THF, 9.1 ml, 18.2 mmol, 1.1 mmol) was added to a cold (-78°) soln. of 8a (6.646 g, 16.596 mmol) in THF (80 ml) over 18 min, and the mixture was stirred for an additional 50 min at 78. A soln. of benzyl bromoacetate (5.3 ml, 33.45 mmol, 2 equiv.) in THF (5 ml) was added during 15 min, and the soln. was stirred for another 3 h at -78° , sat. NH₄Cl (50 ml) was added, and the mixture was warmed to r.t. and diluted with H₂O (100 ml) and Et₂O (250 ml). The org. phase was separated and washed with brine (100 ml). Each H₂O fraction was re-extracted with Et₂O (200 ml). Combined org. phases were dried (MgSO₄) and concentrated in vacuo to yield a yellowish solid, which was washed with pentane/Et₂O 3:1 (3 \times 40 ml) to provide crude 9a containing traces of unreacted benzyl bromoacetate, which were evaporated in vacuo with heating (100°). This procedure led to **9a** (6.965 g, 76%) with 1.7% (w/w) of bromoacetate. FC (pentane/ Et₂O 3:1) yielded anal. pure **9a**. White solid. M.p. 141 – 142°. [α]₁^{tt}</sub> = +63.6 (c = 1.0, CHCl₃). IR (CHCl₃): 3037w, 2970w, 1779s, 1730s, 1454m, 1385s, 1370s, 1309m, 1159s, 1085m, 1017w. ¹ H-NMR (400 MHz, CDCl3): 0.87 $(d, J = 2.0, \text{ Me})$; 0.88 $(d, J = 2.2, \text{ Me})$; 1.66 $(s, t - Bu)$; 2.25 - 2.35 (m, Me_2CH) ; 2.50 $(dd, J = 4.3, 17.1, 1 H$, $C(O)CH₂$); 2.73 (dd, J = 9.1, 14.0, 1 H, indole-CH₂); 2.97 (dd, J = 10.4, 17.1, 1 H, C(O)CH₂); 3.16 (ddd, J = 1.0,

6.0, 14.0, 1 H, indole-CH₂); 4.01 – 4.06 (*m*, 1 H, OCH₂CH); 4.15 (*dd*, *J* = 2.7, 9.1, 1 H, OCH₂CH); 4.33 (*ddd*, *J* = 2.7, 3.7, 8.4, NCH); 4.59–4.68 $(m, C(O)CH)$; $\nu_A = 5.02$, $\nu_B = 5.06$ $(AB, J = 12.3, PhCH₂)$; 7.19–7.35 $(m, 7 \text{ arc})$ om. H); 7.43 (s, 1 arom. H); 7.74 (ddd, J = 0.7, 1.3, 7.7, 1 arom. H); 8.11 (br. d, J = 8.0, arom. NCH). ¹³C-NMR (100 MHz, CDCl₃): 14.5, 18.0 (Me); 27.3 (CH₂); 28.2 (CH); 28.2 (Me); 35.8 (CH₂); 39.7, 58.9 (CH); 63.2, 66.5 (CH₂); 83.6 (C); 115.1 (CH); 116.7 (C); 119.5, 122.7, 124.5, 124.5, 128.2, 128.2, 128.5 (CH); 130.1, 135.5, 135.7, 149.6, 153.6, 171.7, 175.0 (C). HR-MALDI-MS: 571.2421 (+1.05 ppm) $(C_{31}H_{36}N_2O_7Na^+$; calc. 571.2415). Anal. calc. for $C_{31}H_{36}N_2O_7$ (548.63): C 67.87, H 6.61, N 5.11; found: C 67.88, H 6.59, N 5.11.

Benzyl (R)-3-({1-[(tert-Butoxy)carbonyl]-1H-indol-3-yl}methyl)-4-[(S)-4-isopropyl-5,5-diphenyl-2-oxooxazolidin-3-yl]-4-oxobutanoate (9b). Analogously to the preparation of 9a, 8b (534 mg, 0.966 mmol) was reacted. FC (pentane/Et₂O 10:1 to 6:1) yielded **9b** (608 mg, 89%). White solid. M.p. 77–79°. $[a]_D^{\text{rt}} = -102.9$ $(c = 0.75, CHCl₃)$. IR (CHCl₃): 3036m, 2982m, 2933m, 1780s, 1727s, 1606w, 1494w, 1452s, 1370s, 1319m, 1085s, $1053w$, $1019w$, $1002w$, $939w$. ¹H-NMR (400 MHz, CDCl₃): 0.83 (d, $J = 6.7$, Me); 0.89 (d, $J = 7.0$, Me); 1.65 (s, t-Bu); $1.95 - 2.03$ (m, Me_2CH); 2.29 (dd, $J = 10.4$, 14.2 , 1 H, indole-CH₂); 2.39 (dd, $J = 4.1$, 17.2 , 1 H, C(O)CH₂); 2.55 (ddd, $J = 1.0, 4.8, 14.2, 1$ H, indole-CH₂); 2.89 (dd, $J = 10.7, 17.2, 1$ H, C(O)CH₂); 4.51 – 4.61 (m, C(O)CH); $v_A = 4.99$, $v_B = 5.06$ (*AB*, J = 12.3, PhC*H*₂); 5.39 (*d*, J = 3.1, NCH); 7.16 – 7.47 (*m*, 18 arom. H); 7.62 – 7.65 $(m, 1 \text{ arom. H})$; 8.06 (br. $d, J = 7.8$, arom. NCH). ¹³C-NMR (100 MHz, CDCl₃): 16.1, 21.5 (Me); 26.7 (CH₂); 28.2 (Me); 30.0 (CH); 35.2 (CH₂); 39.3, 65.2 (CH); 66.5 (CH₂); 83.5, 89.5 (C); 115.1 (CH); 116.5 (C); 119.5, 122.7, 124.2, 124.4, 125.4, 125.9, 128.0, 128.2, 128.3, 128.4, 128.5, 128.6, 128.6, 128.8 (CH); 130.0, 135.4, 135.7, 138.0, 142.3, 149.5, 153.0, 171.7, 174.7 (C). HR-MALDI-MS: 723.3039 (-0.28 ppm) (C₄₃H₄₄N₂O₇Na⁺; calc. 723.3041). Anal. calc. for C₄₃H₄₄N₂O₇ (700.83): C 73.69, H 6.33, N 4.00; found: C 73.70, H 6.54, N 3.92.

(R)-3-({1-[(tert-Butoxy)carbonyl]-1H-indol-3-yl}methyl)-4-[(S)-4-isopropyl-2-oxooxazolidin-3-yl]-4-oxobutyric Acid (10). The soln. of 9a (6.436 g, 11.731 mmol) in THF (130 ml) was stirred under H₂ (1 atm., ballon) in the presence of 10% Pd/C (510 mg) for 4 h. Pd/C was filtered off, washed with THF, and filtrate was concentrated in vacuo to yield an oil. Et₂O was added and evaporated in vacuo to yield 10 (5.838 g, quant.; pure by ¹H-NMR) as a white solid. An anal. sample was obtained by recrystallization from cyclohexane/AcOEt. M.p. 144 ± 145. IR (CHCl3): 2974m, 1779s, 1726s, 1453s, 1386s, 1360s, 1308m, 1158s, 1085m, 1057w, 1018w, 98 4w, 946w. ¹H-NMR (400 MHz, CDCl₃): 0.85 (d, J = 6.9, Me); 0.88 (d, J = 7.0, Me); 1.66 (s, t-Bu); 2.29 – 2.38 (m, Me₂CH); 2.46 $(dd, J=3.9, 17.8, 1 H, C(O)CH₂)$; 2.70 $(dd, J=9.7, 14.0, 1 H, \text{ indole-CH}₂)$; 2.96 $(dd, J=11.0, 17.8, 1 H,$ $C(O)CH₂$); 3.13 – 3.18 (m, 1 H, indole-CH₂); 4.05 (t, $J = 8.8$, 1 H, CH₂O); 4.16 (dd, $J = 2.7, 9.0, 1$ H, CH₂O); 4.33 (m, NCH) ; 4.53 – 4.61 $(m, C(O)CH)$; 7.22 – 7.27 $(m, 1 \text{ arom. H})$; 7.28 – 7.33 $(m, 1 \text{ arom. H})$; 7.44 $(s, 1 \text{ arom. H})$; 7.73 – 7.76 (*m*, 1 arom. H); 8.11 (br. *d*, *J* = 7.6, arom. NCH). ¹³C-NMR (100 MHz, CDCl₃): 14.1, 17.9 (Me); 27.3 $(CH₂)$; 27.9 (CH); 28.2 (Me); 35.3 (CH₂); 39.3, 58.8 (CH); 63.1 (CH₂); 83.7 (C); 115.2 (CH); 116.4 (C); 119.4, 122.7, 124.5, 124.5 (CH); 130.1, 135.5, 149.6, 153.6, 175.6, 177.6 (C). HR-MALDI-MS: 481.1943 (-0.42 ppm) $(C_{24}H_{30}N_2O_7Na^+$; calc. 481.1945). Anal. calc. for $C_{24}H_{30}N_2O_7$ (458.50): C 62.87, H 6.59, N 6.11; found: C 62.84, H 6.60, N 6.14.

(R)-3-({(3R/S)-1-[(tert-Butoxy)carbonyl]-2,3-dihydro-1H-indol-3-yl}methyl)-4-[(S)-4-isopropyl-2-oxo $oxazolidin-3-yl/4-oxobutyric Acid (11)$. The soln. of 9a (1.116 g, 2.034 mmol) in MeOH (20 ml) and THF (20 ml) was stirred under H₂ (1 atm, balloon) in the presence of 10% Pd/C (113 mg) at 0° to r.t. for 13 h. Pd/C was filtered off, washed with THF and MeOH, and concentrated in vacuo. FC (pentane/Et₂O/AcOH $200:100:1$) yielded 11 (787 mg, 84%) as a mixture of epimers. White solid. ¹³C-NMR (100 MHz, CDCl₃): 14.2, 14.3, 17.9, 18.0 (Me); 27.9, 28.0 (CH); 28.4, 28.5 (Me); 36.6 (CH); 37.0, 43.3, 52.6, 54.2 (CH2); 58.7, 58.8 (CH); 63.0, 63.1, 64.3 (CH2); 81 (C); 114.7, 114.8, 122.1, 122.3, 124.1, 128.0, 128.1, 128.2, 129.0 (CH); 134, 143, 152.5, 153.4, 153.5, 174.9, 175.1, 177.0 (C). HR-MALDI-MS: 483.2095 (-1.45 ppm) (C₂₄H₃₂N₂O₇Na⁺; calc. 483.2102). Anal. calc. for $C_{24}H_{32}N_2O_7$ (460.52): C 62.59, H 7.00, N 6.08; found: C 62.55, H 7.01, N 5.91.

(S)-3-[(2R)-2-({[(Benzyloxy)carbonyl]amino}methyl)-3-{1-[(tert-butoxy)carbonyl]indol-3-yl}-1-oxopropyl]-4-isopropyloxazolidin-2-one (12). a) To a stirred soln. of 10 (419 mg, 0.913 mmol) in toluene (5 ml), Et₃N (0.25 ml, 1.79 mmol, 2 equiv.), diphenylphosphoryl azide $((PhO)₂P(O)N₃; 0.24 ml, 1.114 mmol,$ 1.2 equiv.), and BnOH (0.19 ml, 1.84 mmol, 2 equiv.) were added, and the mixture was stirred for 1 h at r.t., followed by refluxing for additional 1 h. Toluene was evaporated in vacuo, the residue was dissolved in Et_oO (30 ml) and 2M HCl (20 ml). The org. phase was separated, washed with sat. NaHCO₃, dried (MgSO₄), and concentrated in vacuo. FC (CH₂Cl₂/Et₂O 50:1) yielded **12** (380 mg, 74%).

b) To a stirred soln. of the epimeric acids 11 (596 mg, 1.294 mmol) in toluene (6 ml), Et₃N (0.36 ml, 2.58 mmol, 2 equiv.), (PhO)₂P(O)N₃ (0.33 ml, 1.53 mmol, 1.2 equiv.), and BnOH (0.27 ml, 2.61 mmol, 2 equiv.) were added, and the mixture was stirred for 1 h at r.t., followed by refluxing for an additional 1 h. Toluene was evaporated in vacuo, and the residue was dissolved in Et₂O (30 ml) and 2_M HCl (20 ml). The org. phase was separated, washed with sat. NaHCO₃, dried (MgSO₄), and concentrated in vacuo to yield 860 mg of yellow oil.

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FC (CH₂Cl₂/Et₂O 50:1 to 25:1) yielded (S)-3-((2R)-2-{[(Benzyloxycarbonyl)amino]methyl}-3-{1-[(tertbutoxy)carbonyl]-2,3-dihydroindol-3-yl}-1-oxopropyl)-4-isopropyloxazolidin-2-one (493 mg, 67%). HR-MALDI-MS: 588.2681 (+0.17 ppm) ($C_{31}H_{39}N_3O_7Na^+$; calc. 588.2680). This product was dissolved in CH₂Cl₂ (6 ml), and DDQ (207 mg, 0.912 mmol, 1.05 equiv.) was added in 3 portions over 5 min. A black mixture formed and was stirred at r.t. for 1 h, filtered, and concentrated in vacuo. FC (CH₂Cl₂/Et₂O 50:1) yielded **12** (276 mg, 38% over two steps). M.p. 83–84°. α ₁th₁ = +45.5 (c = 0.96, CHCl₃). IR (CHCl₃): 3449w, 3032w, 2975w, 1778s, 1724 s, 1608 w, 1514 m, 1453 s, 1386 s, 1371 s, $1302m$, 1157 s, $1087m$, $1017w$. 1 H-NMR $(500 \text{ MHz}, \text{CDCl}_3)$: 0.79 $(d, J =$ 6.9, Me); 0.86 $(d, J = 7.0, \text{Me})$; 1.66 $(s, t$ -Bu); 2.25 – 2.33 $(m, \text{Me}_2\text{CH})$; 2.87 – 2.95 $(m, 1 \text{ H}, \text{ indole-CH}_2)$; 3.14 $(dd, J=8.1, 14.3, 1 \text{ H}, \text{indole-CH}_2$); $3.50-3.62$ (m, NCH₂); $3.91-3.95$ (m, 1 H, OCH₂CH); 4.09 (dd, $J=2.4, 9.0$, 1 H, OCH₂CH); 4.23–4.26 (m, NCH); 4.43–4.46 (m, C(O)CH); $v_A = 5.06$, $v_B = 5.09$ (AB, J = 12.3, PhCH₂); 5.11 – 5.15 (m, NH) ; 7.20 – 7.32 $(m, 7 \text{ arom. H})$; 7.42 $(s, 1 \text{ arom. H})$; 7.55 $(d, J = 7.6, 1 \text{ arom. H})$; 8.11 (br. $d, J =$ 7.4, arom. NCH). ¹³C-NMR (125 MHz, CDCl₃): 14.6, 18.0 (Me); 25.2 (CH₂); 28.2 (Me); 28.5 (CH); 43.0 (CH₂); 43.5, 58.9 (CH); 63.4, 66.8 (CH₂); 83.6 (C); 115.2 (CH); 116.8 (C); 119.0, 122.6, 124.1, 124.5, 128.1, 128.5 (CH); 130.2, 135.4, 136.5, 149.6, 153.8, 156.3, 174.3 (C). HR-MALDI-MS: 586.2525 (+0.17 ppm) $(C_{31}H_{37}N_3O_7Na^+$; calc. 586.2524).

REFERENCES

- [1] D. Seebach, M. Rueping, P. I. Arvidsson, T. Kimmerlin, P. Micuch, C. Noti, D. Langenegger, D. Hoyer, Helv. Chim. Acta 2001, 84, 3503.
- [2] W. C. Wimley, S. H. White, Nat. Struct. Biol. 1996, 3, 842.
- [3] a) M. Schiffer, C.-H. Chang, F. J. Stevens, Protein Eng. 1992, 5, 213; W.-M. Yau, W. C. Wimley, K. Gawrich, S. H. White, Biochemistry 2001, 40, 5000; A. N. J. A. Ridder, S. Morein, J. G. Stam, A. Kuhn, B. Kruijff, J. A. Killian, Biochemistry 2000, 39, 6521.
- [4] a) W. van't Hof, E. C. I. Veerman, E. J. Helmerhorst, A. V. N. Amerongen, Biol. Chem. 2001, 382, 597; b) N. Sitaram, R. Nagaraj, Biochim. Biophys. Acta 1999, 1462, 29.
- [5] a) A. Giacometti, O. Cirioni, G. Greganti, M. Quarta, G. Scalise, Antimicrob. Agents Chemother. 1998, 42, 3320; b) A. Giacometti, O. Cirioni, F. Barchiesi, M. S. Del Prete, G. Scalise, Peptides 1999, 20, 1265; c) A. Giacometti, O. Cirioni, M. S. Del Prete, A. M. Paggi, M. M. D'Errico, G. Scalise, Peptides 2000, 21, 1155.
- [6] I. Ahmad, W. R. Perkins, D. M. Lupan, M. E. Selsted, A. S. Janoff, Biochim. Biophys. Acta 1995, 1237, 109.
- [7] S. B. Aley, M. Zimmerman, M. Hetsko, M. E. Selsted, F. D. Gillin, Infect. Immun. 1994, 62, 5397.
- [8] B. Yasin, M. Pang, J. S. Turner, Y. Cho, N. N. Dinh, A. J. Waring, R. I. Lehrer, E. A. Wagar, Eur. J. Clin. Microbiol. Infect. Dis. 2000, 19, 187.
- [9] C. Subbalakshmi, V. Krishnakumari, N. Sitaram, R. Nagaraj, J. Bioscience 1998, 23, 9.
- [10] T. J. Falla, D. N. Kurunaratne, R. E. W. Hancock, J. Biol. Chem. 1996, 271, 19298.
- [11] a) C. Subbalakshmi, V. Krishnakumari, R. Nagaraj, N. Sitaram, FEBS Lett. 1996, 395, 48 ; b) C. Subbalakshmi, E. Bikshapathy, N. Sitaram, R. Nagaraj, Biochem. Biophys. Res. Commun. 2000, 274, 714; c) P. Staubitz, A. Peschel, W. F. Nieuwenhuizen, M. Otto, F. Götz, G. Jung, R. W. Jack, J. Pept. Sci. 2001, 7, 552.
- [12] C. Lawyer, S. Pai, M. Watabe, P. Borgia, T. Mashimo, L. Eagleton, K. Watabe, FEBS Lett. 1996, 390, 95.
- [13] D. J. Schibli, P. M. Hwang, H. J. Vogel, *Biochemistry* 1999, 38, 16749.
- [14] a) M. B. Strøm, Ø. Rekdal, J. S. Svedsen, *J. Peptide Res.* 2000, 56, 265; b) B. E. Haug, J. S. Svendsen, *J.* Peptide Sci. 2001, 7, 190; c) B. E. Haug, M. L. Skar, J. S. Svendsen, J. Peptide Sci. 2001, 7, 425; D. J. Schibli, P. M. Hwang, H. J. Vogel, FEBS Lett. 1999, 446, 213.
- [15] a) N. G. Park, J.-K. Seo, H.-J. Ku, S.-H. Kim, S. Lee, G. Sugilhara, K.-H. Kim, J.-S. Park, S.-W. Kang, Bull. Korean Chem. Soc. 1997, 18, 50; b) K. Yu, S. Kang, N. Park, J. Shin, Y. Kim, J. Peptide Res. 2000, 55, 51.
- [16] D. Oh, S. Y. Shin, S. Lee, J. H. Kang, S. D. Kim, P. D. Ryu, K.-S. Hahm, Y. Kim, Biochemistry 2000, 39, 11855.
- [17] J. H. Kang, S. Y. Shin, S. Y. Jang, K. L. Kim, K.-S. Hahm, *Biochem. Biophys. Res. Commun.* 1999, 264, 281.
- [18] a) P. Wentworth, L. H. Jones, A. D. Wentworth, X. Zhu, N. A. Larsen, I. A. Wilson, X. Xu, W. A. Goddard, K. D. Janda, A. Eschenmoser, R. A. Lerner, Science 2001, 293, 1806; b) A. D. Wentworth, L. H. Jones, P. Wentworth, K. D. Janda, R. A. Lerner, Proc. Natl. Acad. Sci. U.S.A. 2000, 97, 10930.
- [19] A. S. Lodokhin, in 'Encyclopedia of Analytical Chemistry: Applications, Theory, and Instrumentation, Ed. R. A. Meyers, John Wiley 2000; pp. 5762 – 5779.

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- [20] J. T. Vivian, P. R. Callis, Biophys. J. 2001, 80, 2093.
- [21] D. Seebach, S. Abele, K. Gademann, G. Guichard, T. Hintermann, B. Jaun, J. L. Matthews, J. V. Schreiber, L. Oberer, U. Hommel, H. Widmer, Helv. Chim. Acta 1998, 81, 932.
- [22] D. Seebach, S. Abele, K. Gademann, B. Jaun, Angew. Chem., Int. Ed. 1999, 38, 1595.
- [23] X. Daura, K. Gademann, H. Schäfer, B. Jaun, D. Seebach, W. F. van Gunsteren, J. Am. Chem. Soc. 2001, 123, 2393.
- [24] D. Seebach, M. Overhand, F. N. M. Kühnle, B. Martinoni, L. Oberer, U. Hommel, H. Widmer, Helv. Chim. Acta 1996, 79, 913.
- [25] J. Frankenpohl, P. I. Arvidsson, J. V. Schreiber, D. Seebach, ChemBioChem. 2001, 2, 445.
- [26] K. Gademann, T. Kimmerlin, D. Hoyer, D. Seebach, J. Med. Chem. 2001, 44, 2460.
- [27] M. Werder, H. Hauser, S. Abele, D. Seebach, Helv. Chim. Acta 1999, 82, 1774.
- [28] a) M. Rueping, Y. Mahajan, M. Sauer, D. Seebach, ChemBioChem 2002, 3, 257; b) N. Umezawa, M. A. Gellman, M. C. Haigis, R. T. Raines, S. H. Gellman, J. Am. Chem. Soc. 2002, 124, 368.
- [29] K. Gademann, M. Ernst, D. Seebach, D. Hoyer, Helv. Chim. Acta 2000, 83, 16.
- [30] a) H. Franzen, L. Grehn, U. Ragnarsson, J. Chem. Soc., Chem. Commun. 1984, 1699; b) 'Synthesis Notes', Eds. B. Dörner and P. White, in Novabiochem Catalog, 2000, Chapt. B 1.7, p. B6-B9.
- [31] a) T. Hintermann, D. Seebach, Helv. Chim. Acta 1998, 81, 2093; b) C. J. Barnett, T. M. Wilson, D. A. Evans, T. C. Somers, Tetrahedron Lett. 1997, 38, 735.
- [32] L. Ruiyan, P. Zhang, T. Gan, J. M. Cook, J. Org. Chem. 1997, 62, 7447.
- [33] a) E. Arvanitis, H. Ernst, A. A. Ludwig, A. J. Robinson, P. B. Wyatt, J. Chem. Soc., Perkin Trans. 1 1998, 521; b) D. A. Evans, L. D. Wu, J. M. Wiener, J. S. Johnson, D. H. B. Ripin, J. S. Tedrow, J. Org. Chem. 1999, 64, 6411; c) M. P. Sibi, P. K. Deshpande, J. Chem. Soc., Perkin Trans. 1 2000, 1461; d) D. Seebach, T. Sifferlen, P. A. Mathieu, A. M. Häne, C. M. Krell, D. J. Bierbaum, S. Abele, Helv. Chim. Acta 2000, 83, 2849
- [34] H. Okubo, F. Feng, D. Nakano, T. Hirata, M. Yamaguchi, T. Miyashita, Tetrahedron 1999, 55, 14855.
- [35] E. v. Hungerbuehler, D. Seebach, D. Wasmuth, *Helv. Chim. Acta* 1981, 64, 1467. [36] a) D. O'Hagan, M. Tavasli, Tetrahedron: Asymmetry 1999, 10, 1189; b) D. J. Bailey, D. O'Hagan, M.
- Tavashi, Tetrahedron: Asymmetry 1997, 8, 149
- [37] G. Guichard, S. Abele, D. Seebach, Helv. Chim. Acta 1998, 81, 187.

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